



REVISION WORKSHEET-GRADE11

SUBJECT: CHEMISTRY

1. Azimuthal quantum number of 3d electron is
(a) 2 (b) 0 (c) 3 (d) +1
2. Which of the following is responsible to rule out the existence of definite paths or trajectories of electrons?
(i) Pauli's exclusion principle.
(ii) Heisenberg's uncertainty principle.
(iii) Hund's rule of maximum multiplicity.
(iv) Aufbau principle
3. Which of the following is not an actinoid?
(i) Curium ($Z = 96$)
(ii) Californium ($Z = 98$)
(iii) Uranium ($Z = 92$)
(iv) Terbium ($Z = 65$)
4. The statement that is not correct for periodic classification of elements is:
(i) The properties of elements are periodic function of their atomic numbers.
(ii) Nonmetallic elements are less in number than metallic elements.
(iii) For transition elements, the 3d-orbitals are filled with electrons after 3p-orbitals and before 4s-orbitals.
(iv) The first ionization enthalpies of elements generally increase with Increase in atomic number as go along a period
5. The pair of ions having same electronic configuration is _____
(i) Cr^{2+} , Fe^{2+} (ii) Fe^{3+} , Mn^{2+}
(iii) Fe^{2+} , Co^{2+} (iv) Sc^{3+} , Cr^{3+}
6. Which of the following is the correct order of increasing tendency to gain electron
(i) $1s^2 2s^2 2p^5$ (ii) $1s^2 2s^2 2p^4$
(iii) $1s^2 2s^2 2p^4 3s^1$ (iv) $1s^2 2s^2 2p^5$

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Xiaolong Qi



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