

**GRADE 10****ACTIVITIES****Activity 1**

1. Give ONE word/term for each of the following descriptions. Write only the word/term next to the question number (1.1 – 1.6).

- 1.1 Process whereby carbon atoms bond with itself to form a long chain.
- 1.2 A group of two or more atoms which are bonded together to form a compound which behaves differently from the original atoms.
- 1.3 The arrangement of elements according to their atomic number.
- 1.4 The chemical substance consisting of one type of atom and cannot be broken down.
- 1.5 The reaction occurs when an isotope of an atom gives off bursts of energy.
- 1.6 The chemical reaction that occurs when water is added to a molecule to break the bonds holding its atoms together.

(1 x 6 = 6)

**Activity 2**

2. Change the **UNDERLINED** word in each of the following statements to make it **TRUE**. Write only the correct word next to the question number (2.1 – 2.5).

- 2.1 A molecule is when two or more substances combine that each retains its own chemical identity.
- 2.2 Ductility is the sharing of electrons by atoms.
- 2.3 Period 7 are elements in the periodic table which have 7 electrons in their outer energy level and are most reactive.
- 2.4 Isotopes are atoms which have the same atomic number but have different mass number.
- 2.5 Electrons are positively charged and are found in the nucleus of an atom.

(1x5 = 5)

**Activity 3**

3. Indicate whether each of the following statement in **COLUMN A** applies to **A ONLY**, **B ONLY**, **BOTH A and B** or **NONE**. Write **A ONLY**, **B ONLY**, **BOTH A and B** or **NONE** next to the question number (3.1 – 3.6).

<b>COLUMN A</b>	<b>COLUMN B</b>
3.1 A. Anion B. Cation	Atoms with more electrons.
3.2 A. Valence electron B. Valence shell	An outermost energy level of an atom.
3.3 A. Atomic number B. atomic mass	Number of protons and neutrons.
3.4 A. Carbon B. Hydrogen	An element which is able to catenate.
3.5 A. Covalent bonding B. Ionic bonding	Copper chloride
3.6 A. Catabolic B. Anabolic	A process that occurs when the molecules or compounds are broken down into simpler substances.

(1x6 = 6)

# Paper 1 Agric Grade 11

**Mitchell Berger**



## Paper 1 Agric Grade 11:

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