Mayya's Resonance Guide



Resonance structures are multiple electronic representations of the same molecule. The only difference between resonance structures is the location of electrons.

Curved Arrows represent the movement of an electron pair



Wrong Curved Arrows

Brandformation



as carbocation, C+

Wrong because hydrogen has no electrons and an arrow is always coming from something that has electrons!



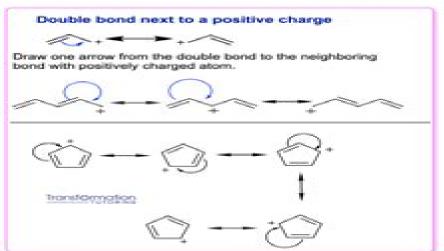
Wrong because the carbon on the right already has an octet and would have 10 electrons

Promoficing-afford

Curved Arrows Rules:

exceed an octet (8 electrons):

Main Resonance Patterns



1.Second row elements (C, N, O.,) CANNOT

An arrow should always start from a place that has electrons such as a lone pair or a bond





something that can accept a pair of electrons, such

Draw one arrow from the lone pair to the neighboring bond and another arrow from the double bond to the atom.

Example:

A lone pair next to a positive charge



Draw one arrow from the lone pair to the neighboring bond with the positively charge atom.

Note: The net charge of all resonance forms must be equal

Hint: Draw out all of the carbons, hydrogens and lone pairs of electrons when you begin to practice resonance

Relative importance of resonance forms

The more bonds in a resonance form, the more important it is.



2. The less charge, the better resonance is.



Transformation

- The more electronegative an atom is, the easier it is for it to bear a negative charge, and not positive charge.
 - 0:> 0:

oxygen is more electronegative than carbon and can bear the negative charge better.

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Resonance Structures Cheat Sheet

Arturo Cuomo

Resonance Structures Cheat Sheet:

Organic Chemistry I For Dummies Arthur Winter, 2016-05-13 Organic Chemistry I For Dummies 2nd Edition 9781119293378 was previously published as Organic Chemistry I For Dummies 2nd Edition 9781118828076 While this version features a new Dummies cover and design the content is the same as the prior release and should not be considered a new or updated product The easy way to take the confusion out of organic chemistry Organic chemistry has a long standing reputation as a difficult course Organic Chemistry I For Dummies takes a simple approach to the topic allowing you to grasp concepts at your own pace This fun easy to understand guide explains the basic principles of organic chemistry in simple terms providing insight into the language of organic chemists the major classes of compounds and top trouble spots You ll also get the nuts and bolts of tackling organic chemistry problems from knowing where to start to spotting sneaky tricks that professors like to incorporate Refreshed example equations New explanations and practical examples that reflect today s teaching methods Fully worked out organic chemistry problems Baffled by benzines Confused by carboxylic acids Here's the help you need in plain English NCLEX-RN Review Made Incredibly Easy ,2004-11-09 Revised to meet the latest Board of Nurse Examiners criteria for the NCLEX RN this book uses the well known Incredibly Easy approach to make NCLEX review effective and enjoyable In a light hearted manner that reduces anxiety and aids retention the book thoroughly reviews every area of nursing adult care psychiatric care maternal neonatal care care of the child leadership and management and law and ethics This edition includes a new chapter on how to prepare for the NCLEX plus 200 alternate format questions and answers added to the appendix and accompanying CD ROM The book also includes an entertaining graphic novel depicting the NCLEX process from application to license and valuable strategies for successfully passing the exam NCLEX-RN® Review Made Incredibly Easy! Lippincott, 2012-02-03 NCLEX RN Review Made Incredibly Easy Fifth Edition uses the well known Incredibly Easy approach to make NCLEX review more interesting and effective Fully revised and updated to match the latest NCLEX RN test plan this title provides solid clinical information and clear rationales written in a conversational easy to understand manner with wealth of unique memory aids to assist with knowledge retention It covers every area of nursing that students will see on the NCLEX exam including adult care pediatric care maternal neonatal care psychiatric care and professional issues Critical thinking questions accompany each chapter and each section and an additional 3 250 challenging NCLEX style practice questions including audio questions and graphic option questions are offered online

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