

Molecular Composition of Gases

SECTION: PART B

QUESTIONS: Answer the following questions in the space provided.

1. _____ The average speed of a gas molecule is most directly related to the _____.
 - (a) volume of the container
 - (b) pressure within the gas
 - (c) temperature of the gas
 - (d) number of molecules in the gas
2. _____ Which value below best represents the average molecular speed of a diatomic gas given at the same temperature?
 - (a) The value associated with nitrogen gas is directly proportional.
 - (b) The value associated with nitrogen gas directly proportional.
 - (c) The value associated with oxygen is half of nitrogen gas directly proportional.
 - (d) The value associated with oxygen is double the nitrogen gas directly proportional.
3. _____ Which combination of conditions corresponds to the following reaction?

$$\text{nitrogen} + \text{oxygen} \rightarrow \text{nitrogen dioxide}$$

 - a. _____ N_2 is present. The volume of O_2 is constant. The volume of N_2O_2 is constant.
 - b. _____ N_2 is present. The mass of O_2 is constant. The mass of N_2O_2 is constant.
4. _____ Which set of 3 samples is at the same pressure?

- a. _____ Each sample contains the same number of molecules.
 - b. _____ Each sample contains the same number of molecules.
 - c. _____ Each sample contains the same number of molecules.

QUESTIONS: Answer the questions in the space provided. Show all your work. Answer all your questions in the space provided.

1. _____ Calculate the volume of 1 mole of gas at 1 atm and 273 K.

Review Molecular Composition Of Gases Short Answer

Arshad Iqbal



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