

Chapter 11 Review Questions

Solutions can be found in Chapter 12.

Section I: Multiple Choice

1. According to the theory put forth by Louis de Broglie, all matter has wave-like properties such as interference, but these properties are only seen at a microscopic scale. Why are these properties not typically observed at a macroscopic scale?
 - (A) The wavelength of matter is typically too large to observe this interference.
 - (B) Interference of matter only occurs when these waves interact with other objects comparable to their wavelength and those things are microscopic.
 - (C) There are no energy level transitions available to allow for this interference to be observed.
 - (D) At the macroscopic scale, the interference is always destructive so it cannot be observed.
2. A metal whose work function is 6.0 eV is struck with light of frequency $7.2 \times 10^{15} \text{ Hz}$. What is the maximum kinetic energy of photoelectrons ejected from the metal's surface?
 - (A) 7 eV
 - (B) 13 eV
 - (C) 19 eV
 - (D) 24 eV
3. An atom with one electron has an ionization energy of 25 eV . How much energy will be released when the electron makes the transition from an excited energy level, where $E = -16 \text{ eV}$, to the ground state?
 - (A) 9 eV
 - (B) 11 eV
 - (C) 16 eV
 - (D) 25 eV
4. The single electron in an atom has an energy of -40 eV when it's in the ground state, and the first excited state for the electron is at -10 eV . What will happen to this electron if the atom is struck by a stream of photons, each of energy 15 eV ?
 - (A) The electron will absorb the energy of one photon and become excited halfway to the first excited state, then quickly return to the ground state, emitting a 15 eV photon in the process.
 - (B) The electron will absorb the energy of one photon and become excited halfway to the first excited state, then quickly absorb the energy of another photon to reach the first excited state.
 - (C) The electron will absorb two photons and be excited to the first excited state.
 - (D) Nothing will happen.
5. The products of several radioactive decays are being studied. Each particle starts with the same speed and enters into a region with a uniform magnetic field directed perpendicular to the initial velocity of the particles. Which observation could be made?
 - (A) A neutron and an electron are deflected in the same direction, but with the electron turning with a larger radius.
 - (B) An alpha particle and an electron are deflected in the same direction, but with the alpha particle turning with a larger radius.
 - (C) An alpha particle and a neutron are both undeflected.
 - (D) An alpha particle is deflected and a neutron is undeflected.

Review Questions Info 1400 Chapter 10

Robert R. Redfield



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Review Questions Info 1400 Chapter 10 Book Review: Unveiling the Power of Words

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Table of Contents Review Questions Info 1400 Chapter 10

1. Understanding the eBook Review Questions Info 1400 Chapter 10
 - The Rise of Digital Reading Review Questions Info 1400 Chapter 10
 - Advantages of eBooks Over Traditional Books
2. Identifying Review Questions Info 1400 Chapter 10
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Review Questions Info 1400 Chapter 10
 - User-Friendly Interface
4. Exploring eBook Recommendations from Review Questions Info 1400 Chapter 10
 - Personalized Recommendations
 - Review Questions Info 1400 Chapter 10 User Reviews and Ratings
 - Review Questions Info 1400 Chapter 10 and Bestseller Lists
5. Accessing Review Questions Info 1400 Chapter 10 Free and Paid eBooks

- Review Questions Info 1400 Chapter 10 Public Domain eBooks
- Review Questions Info 1400 Chapter 10 eBook Subscription Services
- Review Questions Info 1400 Chapter 10 Budget-Friendly Options
- 6. Navigating Review Questions Info 1400 Chapter 10 eBook Formats
 - ePub, PDF, MOBI, and More
 - Review Questions Info 1400 Chapter 10 Compatibility with Devices
 - Review Questions Info 1400 Chapter 10 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Questions Info 1400 Chapter 10
 - Highlighting and Note-Taking Review Questions Info 1400 Chapter 10
 - Interactive Elements Review Questions Info 1400 Chapter 10
- 8. Staying Engaged with Review Questions Info 1400 Chapter 10
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Questions Info 1400 Chapter 10
- 9. Balancing eBooks and Physical Books Review Questions Info 1400 Chapter 10
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Questions Info 1400 Chapter 10
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Questions Info 1400 Chapter 10
 - Setting Reading Goals Review Questions Info 1400 Chapter 10
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Questions Info 1400 Chapter 10
 - Fact-Checking eBook Content of Review Questions Info 1400 Chapter 10
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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