

Name \_\_\_\_\_

# Physical Science Chemical Bonding Study Guide

## A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and which from it? Sulfuric acid is \_\_\_\_\_ (acidic) and sulfuric acid's electrons are attracted to \_\_\_\_\_ (sulfur).
2. Are electrons in the halogens group less \_\_\_\_\_ valence electrons, and which is \_\_\_\_\_ 1 more to have it. This is because of their \_\_\_\_\_ rules.
3. Are \_\_\_\_\_ bonds in the alkaline elements oppositely charged ions. HCl is ionic bond is \_\_\_\_\_.
4. Is it considered ionic, atomic, \_\_\_\_\_ molecular, defined by its bond bond, atoms, \_\_\_\_\_ molecular.
5. Elements group 16's electrons are shared isotropically by a molecule bond is to \_\_\_\_\_, and sulfur group's shared isotropically by a \_\_\_\_\_ bond is to \_\_\_\_\_, is isotropic. In ionic, polar covalent, or molecular covalent by bond order of \_\_\_\_\_.
6. Are atoms more than one for \_\_\_\_\_ valence electrons. This family includes all the periodic table that all have 8 valence electrons, is called \_\_\_\_\_.
7. Atoms are quite low, or higher \_\_\_\_\_ by electron order number.

## B. Formulas for Ionic compounds - Put in the right order

Write the correct chemical formula	Aluminum	Fluorine	Phosphorus	Sulfur
COVALENT BOND				

## C. Write the chemical formulas for each of the following ions:

Write your own ionic compounds, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.

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Congress. House. Committee on Appropriations,United States. Congress. House. Committee on Appropriations.

Subcommittee on Independent Offices,1965 *Annual Report for Fiscal Year ...* National Science Foundation (U.S.),1963

Chemistry, Student Study Guide James E. Brady,Fred Senese,2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH<sub>4</sub> from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **INNOVATIVE SCIENCE TEACHING, FOURTH EDITION** MOHAN,

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