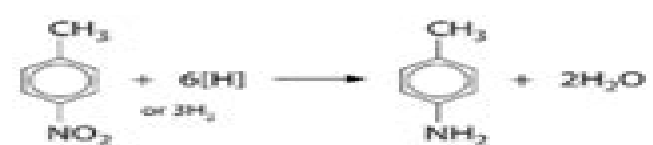


Answers to examination-style questions

Answers	Marks	Examiner's tips
1 a) dimethylamine	1	
b) nucleophilic substitution 	5	The curly arrow must be from the lone pair. Marks 1 and 2 are for the curly arrows if shown (M1 and M2). Mark 3 is for the structure of the intermediate. Mark 4 is to show the removal of the H⁺. You do not need to show the reagent here. Final mark is for the name of the mechanism.
c) quaternary ammonium salt surfactant / detergent / fabric softener or conditioner / hair conditioner	2	
2 a) Sn or Fe/HCl Sn or Fe/H ₂ SO ₄ or H ₂ /Ni can be used as the reducing agent.	2	Do not accept dilute or conc. for these 2 acids or you cannot have HNO₃ at all. Do not accept NaBH₄/ LiAlH₄ or Na/C₂H₅OH.
		
b) 	1	NH ₃ ⁺ is also acceptable.

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A-level Chemistry E. N. Ramsden, 2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study **Calculations for A-level Chemistry** E. N. Ramsden, 1995 Comprehensive mathematics foundation section Work on formulae and equations the mole volumetric analysis and other key areas is included Can be used as a course support book as well as for exam practice Best selling experienced chemistry author

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