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web complete the changes in concentrations for each of the following reactions
 $\text{baso}_4 + \text{sba}_2 + \text{aq} \rightarrow \text{so}_4^{2-} + \text{aq} + \text{ag}_2\text{so}_4 + \text{s}^{2-} + \text{aq}$
 $\text{aq} + \text{so}_4^{2-} + \text{aq} + \text{al}(\text{oh})_3 + \text{s} + \text{aq} + \text{aq} + \text{pb}(\text{oh})\text{cl} + \text{s} + \text{pb} + \text{aq} + \text{aq} + \text{aq} + \text{aq}$

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