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Fission and Fusion

11. The number and mass of nuclei changes in a fission process. The number _____, whereas the mass _____.
12. How is the energy released? How about is the difference between the mass of the separate particles before and the products in a fission reaction?
13. Calculate the energy released in a fission reaction. Consider the energy of two nuclei and neutrons released in the reaction. Consider the splitting of one nucleus and its mass in several pieces (given).
14. Calculate the energy released during the entire synthesis of Helium from protons (assumed to be 100% efficient). 2×10^{26} nuclei of ^1H are there in 1 kg of hydrogen and 10^{26} is 2.0×10^{26} atoms.

$$2.0 \times 10^{26} \times \frac{4.001506 \text{ u}}{4.002602 \text{ u}} = 2.000000 \times 10^{26} \text{ kg}$$

$$2 \times 10^{26} \times (4.001506 \text{ u} - 4.002602 \text{ u}) \times 931 \text{ MeV/u} = 200 \text{ MeV}$$

15. Balance and find the following nuclear reactions. Calculate the energy released in fission.

_____ Mass _____ \rightarrow $^{235}\text{U} \rightarrow$ $^{141}\text{Ba} +$ ^{92}Kr

_____ Mass _____ \rightarrow $^{235}\text{U} +$ $^1_0\text{n} \rightarrow$ ^{141}Ba

_____ Mass _____ \rightarrow $^{235}\text{U} \rightarrow$ $^{141}\text{Ba} +$ $^{92}\text{Kr} +$ $3 \times$ ^1_0n

_____ Mass _____ \rightarrow $^{235}\text{U} \rightarrow$ $^{141}\text{Ba} +$ ^1_0n

_____ Mass _____ \rightarrow $^{235}\text{U} +$ $^1_0\text{n} \rightarrow$ $^{141}\text{Ba} +$ $^{92}\text{Kr} +$ $3 \times$ ^1_0n (then using equation and convert to)

16. a. Calculate the following is an example of a fission reaction?

i. the initial mass

ii. final mass (products)

iii. binding energy given

iv. binding energy given

17. a. The mass of the reaction and mass in a given given reaction controlled by the law

i. mass law conservation

ii. change in conservation of mass

iii. mass energy and binding energy the final state

iv. change in mass

18. Calculate the approximate number of neutrons in fission in fission

Reaction: $^{235}\text{U} +$ $^1_0\text{n} \rightarrow$ $^{141}\text{Ba} +$ $^{92}\text{Kr} +$ $3 \times$ ^1_0n (then using equation and convert to)

Phys Sci Ch 14 Fission And Fusion Worksheet Answers

JE Gale



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