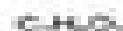


Calculating Percent Composition, Empirical Formulas, and Molecular Formulas

Name: _____

1. Identify the formulas below as either empirical or molecular by writing "EF" for empirical formulas, or "MF" for molecular formulas.



2. Calculate the percent composition of N_2H_4 .

N: _____

H: _____

3. Calculate the percent composition of $\text{C}_2\text{H}_6\text{O}$.

C: _____

H: _____

O: _____

4. How many grams of magnesium, carbon, and oxygen are in a 15g sample of MgCO_3 ?

Mg: _____

C: _____

O: _____

5. How many grams of hydrogen and chlorine are in a 34.1g sample of HClO ?

H: _____

Cl: _____

Percent Composition And Empirical Formulas Answer Key

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**



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