

Topic Modern Periodic Table

Mosley (1914), on the basis of his experiment observed that atomic number is the fundamental property of an element. This concept changed the basis of Periodic Table from atomic weight to atomic number. Mosley proposed his own periodic law, known as Modern Periodic Law. It states "Properties of elements and their compounds are the periodic function of their atomic numbers". Elements were, then, arranged in increasing order of their atomic numbers. The resultant table is known as Modern Periodic Table. In this table sub groups are separated and so the table becomes longer in size. Therefore, it is also known as long form of the periodic table

Main Features of the Modern Periodic Table

It has groups and periods like Mendeleev's Periodic table.

- (ii) It has nine groups, I to VIII and zero group. (Noble gases were known when the table was made)
- (iii) Zero groups contains noble gases (He, Ne, Ar, Kr, Xe and Rn)
- (iv) Groups from I to VII are divided into subgroups A and B.
- (v) Zero group and group VIII has no sub-groups.
- (vi) This table has seven periods.
- (vii) Elements from atomic number 84 (Polonium) onwards are radioactive. A few others having lower atomic number are also radioactive, TC (it is synthetic also).
- (viii) Elements after atomic number 92 (Uranium) are all synthetic or man-made. They have been produced by nuclear reactions.

IUPAC Table

It is simply a changed version of the long form of the periodic table. It was recommended in 1984 by IUPAC committee.

Main Features

- (i) This table has 18 groups, 1 to 18.
- (ii) The notation IA – VIIA, IB – VIIB and VIII has been dropped.
- (iii) Elements of group 1 and group 2 and 13 to 17 have all their inner shells completely filled. These are s- and p- block elements.
- (iv) Elements of group 18 are noble gases. They have all their orbitals completely filled.
- (v) Elements belonging to group 3 to group 12 have their inner (n-1) d or (n-2)f orbital partially filled (these are d- and f-block elements).

Periodic Table Relationship Experiment Chemistry

Xiaolong Qi



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