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Periodic Table Scavenger Hunt

Search your periodic table for the answers to these questions.

Silicon	1. Which element is number 14 on the periodic table?
Cf	2. What is the element symbol for californium?
83	3. How many protons are in an atom of bismuth?
Noble Gas	4. To which element group does argon belong?
Cadmium	5. Which element would you expect to have a higher mass:
	cadmium or zinc?
12.01	. 6. What is the atomic mass of carbon?
Lanthanides	7. What do you call the element series from atomic number 57-71?
Gold	8. Which element has a symbol that starts with a letter different
	from the first one in its name: aluminum, copper, gold,
	rhenium?
Hydrogen	9. Which element has the lowest atomic mass?
Ruthenium	10. What is the first element with an atomic mass greater than 100?
Aluminum	11. What is the first basic metal on the periodic table?
False	12.True or false: Tin and antimony are in the same element group.
Francium	13.What is the heaviest alkali metal?
12	14. How many protons are in an atom of magnesium?
Silicon	15.Which of the following is not a nonmetal: sulfur, oxygen, silicon,
	nitrogen?
Tungsten	16.What is the name of the element with the symbol W?
Palladium	17. Which element has an atomic mass of 106.42?
Halogen	18.Astatine belongs to which element group: nonmetal, halogen, noble gas?
Barium	19.What is the element with the symbol Ba?
J or O	20.Name a letter never used in any element symbol?

Periodic Trends Scavenger Hunt Answer Key

John D. Butler

Periodic Trends Scavenger Hunt Answer Key:

Content Area Reading and Learning Diane Lapp, James Flood, Nancy Farnan, 2016-11-18 How can teachers make content area learning more accessible to their students This text addresses instructional issues and provides a wealth of classroom strategies to help all middle and secondary teachers effectively enable their students to develop both content concepts and strategies for continued learning The goal is to help teachers model through excellent instruction the importance of lifelong content area learning This working textbook provides students maximum interaction with the information strategies and examples presented in each chapter This book is organized around five themes Content Area Reading An Overview The Teacher and the Text The Students The Instructional Program School Culture and Environment in Middle and High School Classrooms Pedagogical features in each chapter include a graphic organizer a chapter overview Think Before Think While and Think After Reading Activities which are designed to integrate students previous knowledge and experience with their new learnings about issues related to content area reading literacy and learning and to serve as catalysts for thinking and discussions This textbook is intended as a primary text for courses on middle and high school content area literacy and learning A Guide for Using the Red Pony in the Classroom Mari Lu Robbins, 1994 Who's the New Kid in Chemistry? John D. Butler, 2013-12-12 Who s the New Kid in Chemistry offers an unprecedented look at student engagement and teacher best practices through the eyes of an educational researcher enrolled as a public high school student Over the course of seventy nine consecutive days John D Butler participates in and observes Rhode Island 2013 Teacher of the Year Jessica M Waters s high school chemistry class documenting his experiences as they unfold Who s the New Kid in Chemistry is a compelling example of what can be accomplished when an educational researcher and teacher collaborate in the classroom This work includes a discussion on flexible homework assignments data driven instruction and thirty teacher best practices This book is an invaluable resource for teachers across all content areas masters and doctoral research method Answer Key for The Mystery of the Periodic Table Staff of Christ the King classes and future Teachers of the Year Books, 2018-02

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