

Periodic Table Hopscotch Laboratory: Dr. Leslie Pierce

Students' Guide

Goals

- Visualize the concept of the mole
- Gain experience in calculating grams and moles

The Activity

In this activity, you will visualize the concept of the mole using a mole of chalk as a model. You will practice calculations of moles and grams, and end up with a better understanding of what a mole is and how chemists use it.

Materials for Each Group

- Data book and pencil
- A chunk of chalk
- Semi-analytical balance
- Electronic balance, if available
- A table of atomic masses
- A calculator

SAFETY

No special safety considerations are required.

Instructions

1. Take a chunk of chalk and measure its mass on the balance: _____
2. Go outside and draw something on the sidewalk.
3. Go back in class and weigh the unused chalk: _____
4. Based on the initial mass of the chalk and the mass at the end, calculate how many grams of chalk you left out on the sidewalk: _____
5. Write down the molecular formula of the chalk: _____
6. Get the masses of the atoms: _____
7. Calculate the chalk's molar mass: _____
8. Using this data, calculate:
 - a. How many moles of chalk did you leave on the sidewalk? _____
 - b. How many molecules of chalk did you leave on the sidewalk? _____
 - c. How many atoms of calcium did you use? _____
 - d. How many atoms of carbon did you use? _____
 - e. How many atoms of oxygen did you use? _____

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Table of Contents Periodic Table Hopscotch Laboratory Answers

1. Understanding the eBook Periodic Table Hopscotch Laboratory Answers
 - The Rise of Digital Reading Periodic Table Hopscotch Laboratory Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Hopscotch Laboratory Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Hopscotch Laboratory Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Hopscotch Laboratory Answers
 - Personalized Recommendations
 - Periodic Table Hopscotch Laboratory Answers User Reviews and Ratings
 - Periodic Table Hopscotch Laboratory Answers and Bestseller Lists
5. Accessing Periodic Table Hopscotch Laboratory Answers Free and Paid eBooks
 - Periodic Table Hopscotch Laboratory Answers Public Domain eBooks
 - Periodic Table Hopscotch Laboratory Answers eBook Subscription Services
 - Periodic Table Hopscotch Laboratory Answers Budget-Friendly Options
6. Navigating Periodic Table Hopscotch Laboratory Answers eBook Formats

- ePub, PDF, MOBI, and More
- Periodic Table Hopscotch Laboratory Answers Compatibility with Devices
- Periodic Table Hopscotch Laboratory Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Hopscotch Laboratory Answers
 - Highlighting and Note-Taking Periodic Table Hopscotch Laboratory Answers
 - Interactive Elements Periodic Table Hopscotch Laboratory Answers
- 8. Staying Engaged with Periodic Table Hopscotch Laboratory Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Hopscotch Laboratory Answers
- 9. Balancing eBooks and Physical Books Periodic Table Hopscotch Laboratory Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Hopscotch Laboratory Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Hopscotch Laboratory Answers
 - Setting Reading Goals Periodic Table Hopscotch Laboratory Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Hopscotch Laboratory Answers
 - Fact-Checking eBook Content of Periodic Table Hopscotch Laboratory Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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