Unit 2: The Periodic Table Review Sheet

UZLI: PT History

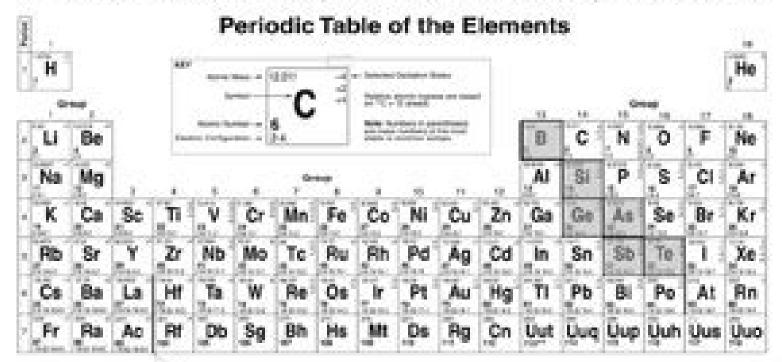
- On the present day Periodic Table, elements are arranged in order of increasing atomic number
- Groups: vertical columns on the periodic table
 - Elements in the same group have similar chemical properties because they have the same number of valence electrons
 - Valence electron: outermost electrons (lost # in the electron configuration)
- Periods: horizontal rows on the periodic table.

U21.2: Ports of the Periodic Table

Elements are classified as either metals, nanmetals, or metalloids.

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Classification can be determined by location on the periodic table or by different properties.



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Periodic Table Review Pt 2

J Rink

Periodic Table Review Pt 2:

Unveiling the Magic of Words: A Overview of "Periodic Table Review Pt 2"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Periodic Table Review Pt 2**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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