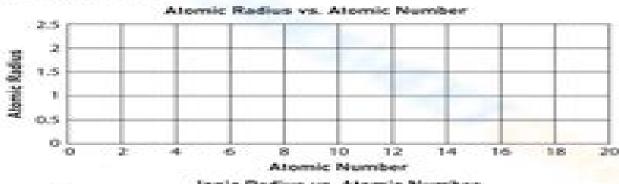
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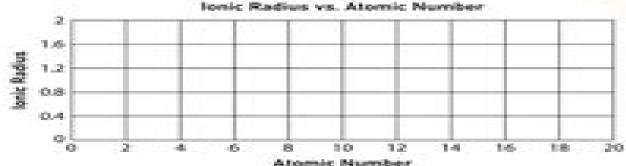
The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	Hile	1.7	Be		C	140	0	- 15
Atomic Number	1	- 2	3	4	5	6	7	. 8	9
Atomic Radius (A)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
Radius (A)	0.012		0.76	0.35	0.23	0.16	0.13	3,4	1.33

740	Péa	Migr	All .	54	8	. 5	CI	. Ast
10	1.11	1.2	13	14	15	16	11.7"	1.6
0:38	2.23	1.22	1.82	1,46	1.23	11.09	0.97	0.21
	1.02	0.72	0.535	05.46	0.38	0.37	1.81	100

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.





<u>Periodic Trends Worksheet Using The Periodic Law</u> <u>Answers</u>

G Thomas

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relevant to cultivating global awareness in primary and secondary students Rather than seeing global education as a special field separate from the other disciplines the author encourages integration of global perspectives into everything we do Showcasing how global awareness is a developmental issue dependent upon the student's ability to step outside of their own place based comfort zone this volume lays out a roadmap of major challenges and issues around instilling this awareness in students This book connects theory and practice about global education relevant to cultivating global awareness in primary and secondary students From this foundation the book engages with the challenge of integrating global perspectives within a crowded curriculum By convincing students and teachers alike of global education s centrality thinking globally becomes an integral component of learning across subject areas and grade levels and this work encourages students to exercise empathy for the other and to develop critical skills to see through media distortions and fake news so they can better resist the tendency of politicians in our increasingly multicultural countries to divide people along racial and ethnic lines Chemistry Robert G. Mortimer, 2008-05-29 In this third edition core applications have been added along with more recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics as well as in the experimental study of extremely rapid chemical reactions Fully revised concise edition covering recent developments in the field Supports student learning with step by step explanation of fundamental principles an appropriate level of math rigor and pedagogical tools to aid comprehension Encourages readers to apply theory in practical situations Multimedia Bookseller ,1996

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