Chapter 1: Intro to earth science

- Geology- The study of earth
- Oceanography- study of the ocean
- Meteorology Study of the atmosphere and weather
- Astronomy-Study of the universe

1. What science is not used within the Earth sciences? Chemistry, Bio, Physics, and mathematics are all used

- Scientific method: Making observation/collect data, develop a hypothesis, conduct experiment, Re-examing the hypothesis and accept or reject it. Make conclusion
- 3. Hypothesis: Untested explanation
- 4. Theory- Tested and confirmed hypothesis

A THEORY is a well-tested and widely accepted view that best explains certain scientific observations

- Paradigm A theory that explains a large number of interrelated aspects of the natural world
- Hydrosphere Includes all water on earth (fresh water found in streams, lakes, and glaciers)
- Atmosphere: The envelope of gases surrounding the earth, thin blanket of air, over 3.5 miles
- 8. Biosphere-includes all life
- Geosphere-Solid earth, extends from the surface to the center of the planet, LARGEST of Earth's 4 spheres, consists of the crust, mantle, and core
 - · Crust-thin, rocky outer skin. Less dense, many rock types
 - Oceanic crust- dense, basaltic rock
 - Mantle- more than 82% of earths volume.
 - Upper mantle- Lithosphere: entire crust and uppermost mantle.
 Asthenosphere: soft layer
 - . Lower mantle- VERY HOT strengthens with depth
 - Outer Core: Liquid layer, earth's magnetic field
 - Inner core-solid

10. Soil belongs to the Geosphere, hydrosphere, atmosphere, and biosphere

- 11. The Crust is the thinnest layer of the earth
- 12. What is the source of the energy that powers the earth system? The sun and heat from earth's interior

Chapter 2: Matter and minerals

- Definition of a mineral: Natural, inorganic, solid, possess an orderly internal structure
 of atoms, have a definite chemical composition
- Rock- and naturally occurring solid mass of mineral or mineral like matter
- · Atoms smallest particles of matter
 - Nucleus- central part of an atom that contains, protons +, neutrons neutral charge
 - (2) Electron shells- surrounds nucleus, contains electrons charges

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O. Ray Whittington

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