

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. The only sure evidence for a chemical reaction is
a. the formation of a gas.
b. changes in properties.
c. the production of one or more new substances.
- _____ 2. When a chemical reaction takes place in an open system,
a. matter cannot move at all.
b. matter can enter from or escape to the surroundings.
c. matter is not allowed to enter from or escape to the surroundings.
- _____ 3. Which of the following is a balanced chemical equation?
a. $2 \text{Fe}_2\text{O}_3 + 3 \text{C} \rightarrow 4 \text{Fe} + 3 \text{CO}_2$
b. $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$
c. $\text{SO}_2 + \text{O}_2 + 2 \text{H}_2\text{O} \rightarrow 4 \text{H}_2\text{SO}_4$
- _____ 4. What type of reaction is $\text{FeS} + 2 \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2\text{S}$?
a. a single replacement reaction
b. a double replacement reaction
c. a synthesis reaction
- _____ 5. What is the best form of fire safety?
a. fire prevention
b. the fire department
c. an extinguisher
- _____ 6. Anything that has mass and takes up space is considered
a. energy.
b. chemistry.
c. matter.
- _____ 7. The best way to put out a small kitchen fire is to
a. blow it out.
b. turn off the appliances.
c. cover it or use a home fire extinguisher.
- _____ 8. A chemical reaction that absorbs energy in the form of heat is described as
a. combustion.
b. unbalanced.
c. endothermic.

Section 11 1 Describing Chemical Reactions Practice Problems Answers

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A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape with a gradient, partially overlapping the bar.

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Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features *CliffsNotes Chemistry Practice Pack* Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level **New KS3 Science Complete**

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Texas State Journal of Medicine, 1906

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Pharmaceutical Journal; ,1908

The Pacific Coast Journal of Homeopathy, 1904

The Boston Medical and Surgical Journal, 1880

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Workbook for Organic Chemistry Jerry Jenkins, 2009-12-25 With authors who are both accomplished researchers and educators Vollhardt and Schore's Organic Chemistry is proven effective for making contemporary organic chemistry accessible introducing cutting edge research in a fresh student friendly way A wealth of unique study tools help students organize and understand the substantial information presented in this course And in the sixth edition the themes of understanding reactivity mechanisms and synthetic analysis to apply chemical concepts to realistic situations has been strengthened New applications of organic chemistry in the life sciences industrial practices green chemistry and environmental monitoring and clean up are incorporated This edition includes more than 100 new or substantially revised problems including new problems on synthesis and green chemistry and new challenging problems

Chemical Interactions, 2005

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