

## 11

## CHEMICAL REACTIONS

## Practice Problems

*In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter 1.*

## SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
4. Write a balanced equation for the production of HCl gas from its elements.
5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



7. Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
8. Write a balanced equation from this word equation.  
aqueous silver nitrate + copper metal  $\rightarrow$  silver metal + aqueous copper nitrate
9. Write a balanced equation for the following word equation.



## SECTION 11.2 TYPES OF CHEMICAL REACTIONS

1. Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
2. Write the balanced equation for the reaction that occurs between aluminum and fluorine.
3. Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
4. Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
5. Write the balanced equation for the combustion of propane ( $\text{C}_3\text{H}_8$ ) to produce carbon dioxide and water vapor.
6. Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

# Section 11 1 Describing Chemical Reactions Workbook

**Kevin T. Patton, Gary A. Thibodeau**



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**Introduction to Chemical Reaction Engineering and Kinetics** Mr. Rohit Manglik, 2024-01-22 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels , **Chemical Reactions** Carol Baldwin, 2004 Materials react together sometimes with explosive results This book tells you everything you need to know about chemical reactions There are loads of photos and facts This will help you to get to grips with the topic and find answers quickly Includes exiting photos bite sized chunks of information and tips for further research that will help with your homework

*Anatomy & Physiology (includes A&P Online course) E-Book* Kevin T. Patton, 2018-01-31 Anatomy Physiology includes A P Online course E Book *X-ray Nanochemistry* Ting Guo, 2018-06-01 This book describes the latest developments in the new research discipline of X ray nanochemistry which uses nanomaterials to enhance the effectiveness of X ray irradiation Nanomaterials now can be synthesized in such a way as to meet the demand for complex functions that enhance the X ray effect Innovative methods of delivering the X rays which can interact with those nanomaterials much more strongly than energetic electrons and gamma rays also create new opportunities to enhance the X ray effect As a result new concepts are conceived and new developments are made in the last decade which are discussed and summarized in this book This book will help define the discipline and encourage more students and scientists to work in this discipline These efforts will eventually lead to formation of a full set of physical chemical and materials principles for this new research field

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