

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of a solution and the pH of a solution based on concentration.
2. Calculate pH and concentration of acids and bases and the pH of a solution based on concentration.
3. Calculate the pH of a solution (20 minutes).

Vocabulary

- | | | |
|-------------------|--------------------------------------|------------------|
| 1. autoionization | 2. autoionization constant (K_w) | 3. acid solution |
| 4. autoionization | 5. autoionization | 6. pH |

Key Equation

- | | |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

Part A Completion

Write the equation and/or the numerical value(s) of the constant(s) that are indicated in the space(s) provided. Write the equation and/or the numerical value(s) in the space(s) provided.

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Section 19 2 Hydrogen Ions And Acidity Practice Problems

Sebastian Brünink



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