

## 7-1 Review and Reinforcement

### Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative  
positive  
electron

noble gas  
atoms  
ion

cation  
anion  
electron

electron configuration  
polyatomic  
empirical formula

1. An ionic bond is an attraction between oppositely charged \_\_\_\_\_. **ions**
2. Anionic bond is \_\_\_\_\_. **negative** change.
3. An ionic bond is formed by losing or gaining \_\_\_\_\_. **electrons**
4. The \_\_\_\_\_ noble gases that cannot bond to gain, lose, or share electrons to obtain to complete a full set of valence electrons.
5. When sodium and chlorine form a bond, both have negative the electron configuration of \_\_\_\_\_ noble gas.
6. The atoms that make up \_\_\_\_\_ \_\_\_\_\_ ions are bonded together by ionic bonds.
7. The \_\_\_\_\_ method can be used to write the formula for an ionic compound.
8. The \_\_\_\_\_ empirical formula is a compound denotes the ratio of ions in the compound. If the element is iron, write "Fe" if it is below. Change the subscript until an equals to make it true. Write your answer on the line.

\_\_\_\_\_ **losing**

\_\_\_\_\_ **iron**

\_\_\_\_\_ **iron**

\_\_\_\_\_ **polyatomic**

\_\_\_\_\_ **anions**

\_\_\_\_\_ **iron (II) oxide**

9. Calcium boride is chemically written as \_\_\_\_\_ **gaining** two electrons.

10. A cation has a \_\_\_\_\_ **positive** charge.

11. A binary ionic compound contains only \_\_\_\_\_ **one** kind of cation and one kind of anion.

12. Substances that consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six \_\_\_\_\_ **electrons**.

14. A great deal of energy is \_\_\_\_\_ **gained** when an ionic compound is formed.

# Section 1 Reinforcement Motion Answers

**Minjie Lin**



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