

Name:	
Grade & Section:	
Subject:	

Nuclear Decay

Types of Radiation

1. Complete the table below to compare the properties of alpha, beta, positron, and gamma radiation.

Name	Alpha	Beta	Positron	Gamma
Greek Letter				
Symbol				
Charge				
Stopped by				

Nuclear Decay Reactions

In problem 2-5, fill in the missing particle for each nuclear decay reaction. Then, match each equation to the appropriate type of nuclear decay.

- A. alpha emission
- B. beta emission
- C. electron capture
- D. positron emission



In problem 6 and 7, write a balanced equation for each for each nuclear decay reaction.

6. Decay of Radium-223 by alpha emission



7. Decay of Iodine-126 by positron emission



8. How are the atomic number and mass number of a nuclide affected by positron emission?
9. How are the atomic number and mass number of a nuclide affected by the emission of gamma radiation?
10. Which has a slightly greater mass in a nuclear reaction, the reactants or the product? Why?

Section 25 2 Concept Review Nuclear Decay

Rosina Ehmann



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Table of Contents Section 25 2 Concept Review Nuclear Decay

1. Understanding the eBook Section 25 2 Concept Review Nuclear Decay
 - The Rise of Digital Reading Section 25 2 Concept Review Nuclear Decay
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 25 2 Concept Review Nuclear Decay
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 25 2 Concept Review Nuclear Decay
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 25 2 Concept Review Nuclear Decay
 - Personalized Recommendations
 - Section 25 2 Concept Review Nuclear Decay User Reviews and Ratings
 - Section 25 2 Concept Review Nuclear Decay and Bestseller Lists
5. Accessing Section 25 2 Concept Review Nuclear Decay Free and Paid eBooks

- Section 25 2 Concept Review Nuclear Decay Public Domain eBooks
- Section 25 2 Concept Review Nuclear Decay eBook Subscription Services
- Section 25 2 Concept Review Nuclear Decay Budget-Friendly Options
- 6. Navigating Section 25 2 Concept Review Nuclear Decay eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 25 2 Concept Review Nuclear Decay Compatibility with Devices
 - Section 25 2 Concept Review Nuclear Decay Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 25 2 Concept Review Nuclear Decay
 - Highlighting and Note-Taking Section 25 2 Concept Review Nuclear Decay
 - Interactive Elements Section 25 2 Concept Review Nuclear Decay
- 8. Staying Engaged with Section 25 2 Concept Review Nuclear Decay
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 25 2 Concept Review Nuclear Decay
- 9. Balancing eBooks and Physical Books Section 25 2 Concept Review Nuclear Decay
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 25 2 Concept Review Nuclear Decay
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 25 2 Concept Review Nuclear Decay
 - Setting Reading Goals Section 25 2 Concept Review Nuclear Decay
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 25 2 Concept Review Nuclear Decay
 - Fact-Checking eBook Content of Section 25 2 Concept Review Nuclear Decay
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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