

Chemistry: Periodic Table Unit Study Guide

Name _____ Date _____ Period _____
Periodic Table Unit Study Guide

Basic Structure and Organization of the Periodic Table
 1. Label each element according to the groups and periods in the periodic table.

- The columns are called **groups**.
- The rows are called **periods**.
- The elements are arranged in order of increasing atomic number.

2. Color & Label the following families on the periodic table below: **Alkali Metals, Alkaline Earth Metals, Transition Metals, Halogens, and Noble Gases**. Write down the atomic number, element name, and the number of protons and neutrons.

Alkali Metals

Alkaline Earth Metals

Transition Metals

Halogens

Noble Gases

3. Rank the periodic families above in order of increasing atomic number.

4. Fill in the following chart:

Element	Astatine	Ca^{2+}	$1s^2 2s^2 2p^6 3s^2$	
Number of Electrons				
Atomic Number (Z)				
Group Number (IUPAC)				
Period Number				

Periodic Table And Chemistry Study Guide Answers

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Periodic Table And Chemistry Study Guide Answers:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011

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Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it

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emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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