

# Periodic Table Basics

**Step 1: Complete the squares for each element by adding the atomic number, name, and atomic mass.**

Write the atomic number at the top of the square.

Write the element's name under the symbol.

Write the atomic mass at the bottom of the square.

**Step 2: Determine the number of protons, neutrons, and electrons in each element.**

**Step 3: Create a Bohr diagram for each element.**

**Step 4: Draw the Lewis Structure for each element.**

The sample card for Boron (B) contains the following information:

- Element Information:** Atomic number 5, Symbol B, Name Boron, Atomic mass 10.81.
- Bohr Diagram:** A diagram showing a nucleus with 5 protons and 5 neutrons, and 5 electrons arranged in two shells (2 in the first shell, 3 in the second shell).
- Lewis Structure:** A diagram showing the element symbol B with three valence electrons represented by dots.
- Protons, Neutrons, and Electrons:** P = 5, N = 6, E = 5.

**Step 5: Use the following colors to shade in the square for each element. You should ONLY color in the small square in the upper left-hand corner and not the entire card.**

Green = Li & Na

Pink = O & S

Blue = Be & Mg

Purple = F & Cl

Orange = B & Al

Red = C & Si

Tan = N & P

Yellow = He, Ne, & Ar

**Step 6: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.**



**Step 7: Answer the questions on the back of this worksheet using the information on your Periodic Table.**

# Periodic Table Basics Answer Key

**Kaplan Test Prep**



## **Periodic Table Basics Answer Key:**

Basics of Analytical Chemistry and Chemical Equilibria Brian M. Tissue, 2013-07-22 Enables students to progressively build and apply new skills and knowledge Designed to be completed in one semester this text enables students to fully grasp and apply the core concepts of analytical chemistry and aqueous chemical equilibria Moreover the text enables readers to master common instrumental methods to perform a broad range of quantitative analyses Author Brian Tissue has written and structured the text so that readers progressively build their knowledge beginning with the most fundamental concepts and then continually applying these concepts as they advance to more sophisticated theories and applications Basics of Analytical Chemistry and Chemical Equilibria is clearly written and easy to follow with plenty of examples to help readers better understand both concepts and applications In addition there are several pedagogical features that enhance the learning experience including Emphasis on correct IUPAC terminology You Try It spreadsheets throughout the text challenging readers to apply their newfound knowledge and skills Online tutorials to build readers skills and assist them in working with the text s spreadsheets Links to analytical methods and instrument suppliers Figures illustrating principles of analytical chemistry and chemical equilibria End of chapter exercises Basics of Analytical Chemistry and Chemical Equilibria is written for undergraduate students who have completed a basic course in general chemistry In addition to chemistry students this text provides an essential foundation in analytical chemistry needed by students and practitioners in biochemistry environmental science chemical engineering materials science nutrition agriculture and the life sciences      CBSE - ICSE Chemistry Part I Chandan Sengupta, First Publication October 2021 Place of Publication Arabinda Nagar Bankura 722101 This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry Some of the worksheets are prepared along with model answers Some other worksheets are meant for self assessment and evaluation purposes It is also observed that some of the topics are specific to the referred curriculum Some other topics are varyingly incorporated in other streams of study Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations      Guide to RRB Junior Engineer Stage II Exam - Physics, Chemistry, General Awareness, Basics of Computers, Environment & Pollution Control Disha Experts, 2019-01-25 The book Guide to RRB Junior Engineer Stage II Online Exam has 4 sections common to all streams General Awareness Physics Chemistry Basics of Computers and Applications Basics of Environment and Pollution Control Each section is further divided into chapters which contains theory explaining the concepts involved followed by MCQ exercises The book provides the past 2014 2015 Solved Questions The detailed solutions to all the questions are provided at the end of each chapter      **Kaplan SAT Subject Test Chemistry 2015-2016** Kaplan Test Prep, 2015-03-03 Essential strategies practice and review to ace the SAT Subject Test Chemistry Getting into a top college has never been

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medicine and radiology However at this point interest in the field outpaces the academic and educational infrastructure needed to train radiopharmaceutical chemists For example the vast majority of texts that address radiopharmaceutical chemistry do so only peripherally focusing instead on nuclear chemistry i e nuclear reactions in reactors heavy element radiochemistry i e the decomposition of radioactive waste or solely on the clinical applications of radiopharmaceuticals e g the use of PET tracers in oncology This text fills that gap by focusing on the chemistry of radiopharmaceuticals with key coverage of how that knowledge translates to the development of diagnostic and therapeutic radiopharmaceuticals for the clinic The text is divided into three overarching sections First Principles Radiochemistry and Special Topics The first is a general overview covering fundamental and broad issues like The Production of Radionuclides and Basics of Radiochemistry The second section is the main focus of the book In this section each chapter s author will delve much deeper into the subject matter covering both well established and state of the art techniques in radiopharmaceutical chemistry This section will be divided according to radionuclide and will include chapters on radiolabeling methods using all of the common nuclides employed in radiopharmaceuticals including four chapters on the ubiquitously used fluorine 18 and a Best of the Rest chapter to cover emerging radionuclides Finally the third section of the book is dedicated to special topics with important information for radiochemists including Bioconjugation Methods Click Chemistry in Radiochemistry and Radiochemical Instrumentation This is an ideal educational guide for nuclear medicine physicians radiologists and radiopharmaceutical chemists as well as residents and trainees in all of these areas     ENC Focus ,2000     *Chemical Principles* Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses *Chemical Principles* helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of *Chemical Principles* is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding     *CCNP BSCI Exam Certification Guide* Clare Gough,2004 Prepare for the CCNP 642 801 BSCI exam with the only Cisco Systems authorized self study preparation book Master advanced routing techniques and practices and prepare for the new exam with the 200 plus question testing engine on CD ROM

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