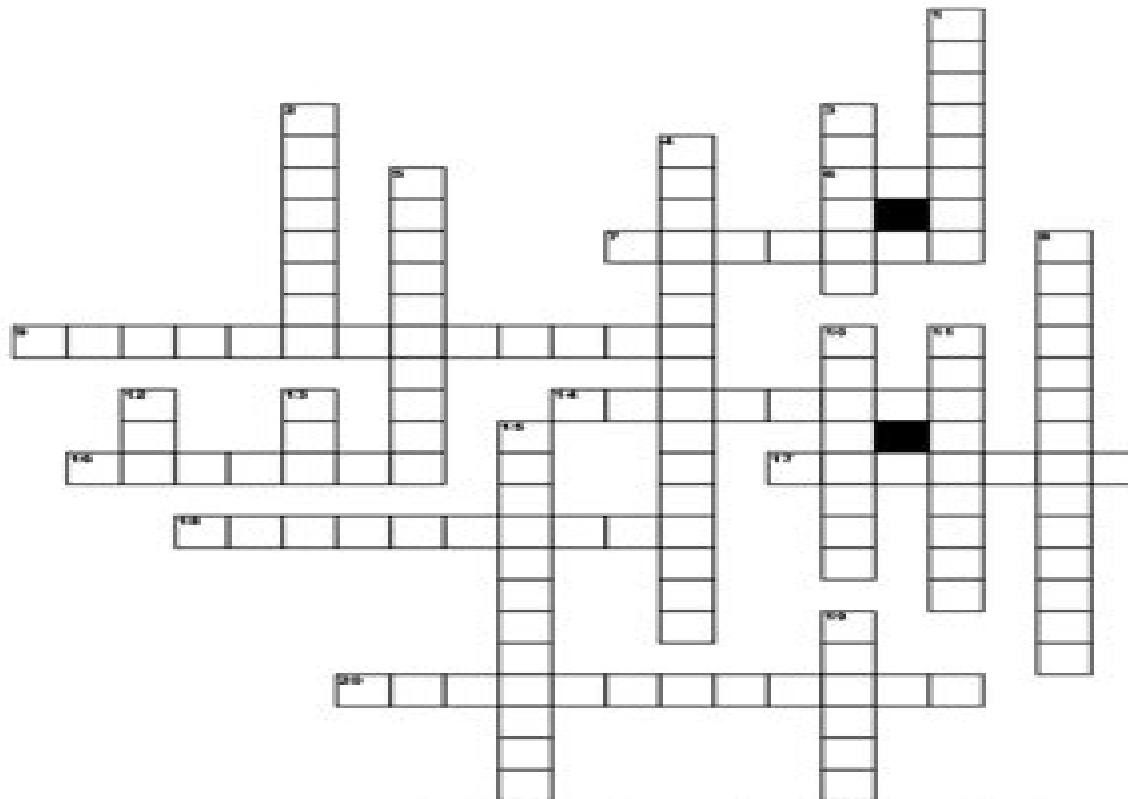


# Structure of the Atom



## Across

6. Mass Number - Atomic Number = Number of Neutrons  
 7. The horizontal rows in the periodic table  
 9. Arrangement of elements according to atomic number on a table  
 14. A single atom or several atoms bound together electro magnetically bonded.  
 16. Center of atom  
 17. Positive particles around the nucleus  
 18. The sum of the number of neutrons and protons

20. 1. All matter consists of minuscule particles called atoms.  
 2. All atoms of a given element are identical to each other. 3. All atoms of a given element are different than those of other elements. 4. Atoms of one element combine with other elements to create compounds. They always combine in equal amounts.

## Down

1. Pure substances that can't be broken down or changed  
 2. 2 or more elements chemically bonded  
 3. Abbreviation for each element  
 4. The electrons on the very outer ring of the atom

5. The property of an atom that causes it to have weight  
 8. Located outside the nucleus with electrons  
 10. Neutral charged particles around the nucleus  
 11. Negative particles outside the nucleus  
 12. Atomic Mass Unit  
 13. Atomic Number = Protons and Electrons  
 15. Quantity of protons and electrons in the nucleus of an atom  
 19. The vertical rows in the periodic table

# Periodic Table Word Search Answers Reinforcement 3

**Abraham**



### **Periodic Table Word Search Answers Reinforcement 3:**

## Reviewing **Periodic Table Word Search Answers Reinforcement 3**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Periodic Table Word Search Answers Reinforcement 3**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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