

CHEMISTRY MULTIPLE CHOICE QUESTIONS

I. Periodicity

2002 -2019

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1. Which species represented by the following formulae has the largest radius?

- A. P^{3-} B. Cl^- C. Ar D. K^+
[2002 M/1 (1.2)]

2. Which of the following oxides is unlikely to dissolve in aqueous sodium hydroxide?

- A. Al_2O_3 B. MgO C. SO_2 D. SiO_2
[2002 M/1 (1.2)]

3. An element of the third period (A to D) is heated in chlorine. The product is purified and then added to water. The resulting solution is found to be neutral.

What is the element?

- A. sodium
B. aluminium
C. silicon
D. phosphorus

[2002 M/1 (1.4)]

4. The chloride of element Q is hydrolysed by water to form an acidic solution and its oxide reacts with acid to form a salt.

What could be the element Q?

- A. magnesium
B. aluminium
C. silicon
D. phosphorus

[2002 Q/1 (1.3)]

5. Which diagram represents the change in ionic radius of the elements across the third period (Na to Cl)?



[2002 Q/1 (1.4)]

6. Which isotope of an element in the third period of the Periodic Table contains the same number of neutrons as ^{32}S ?

- A. ^{24}Mg
B. ^{27}Al
C. ^{28}Si
D. ^{31}P

[2005 M/1 (1)]

7. The successive ionization energies, in $kJ\ mol^{-1}$, of an element X are given below.

$kJ\ mol^{-1}$ 1800 3000 3800 5800 7000 13,000

What is X?

- A. $_{11}Na$ B. $_{12}Mg$ C. $_{13}Al$ D. $_{14}Si$

[2005 M/1 (1.4)]

Periodicity Multiple Choice Questions

M Lipman



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Table of Contents Periodicity Multiple Choice Questions

1. Understanding the eBook Periodicity Multiple Choice Questions
 - The Rise of Digital Reading Periodicity Multiple Choice Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodicity Multiple Choice Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodicity Multiple Choice Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodicity Multiple Choice Questions
 - Personalized Recommendations
 - Periodicity Multiple Choice Questions User Reviews and Ratings
 - Periodicity Multiple Choice Questions and Bestseller Lists

5. Accessing Periodicity Multiple Choice Questions Free and Paid eBooks
 - Periodicity Multiple Choice Questions Public Domain eBooks
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 - Periodicity Multiple Choice Questions Budget-Friendly Options
6. Navigating Periodicity Multiple Choice Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodicity Multiple Choice Questions Compatibility with Devices
 - Periodicity Multiple Choice Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodicity Multiple Choice Questions
 - Highlighting and Note-Taking Periodicity Multiple Choice Questions
 - Interactive Elements Periodicity Multiple Choice Questions
8. Staying Engaged with Periodicity Multiple Choice Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodicity Multiple Choice Questions
9. Balancing eBooks and Physical Books Periodicity Multiple Choice Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodicity Multiple Choice Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodicity Multiple Choice Questions
 - Setting Reading Goals Periodicity Multiple Choice Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodicity Multiple Choice Questions
 - Fact-Checking eBook Content of Periodicity Multiple Choice Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

-
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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