Gizmos

# Lab 22: Student Exploration: Reaction Energy

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: calorimeter, chemical bond, endothermic, enthalpy, exothermic, Hess's law

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Two magnets are stuck together. What might you have to do to get them to separate?

Pull them apart

Suppose you held two magnets a short distance apart, then let go. What would happen?

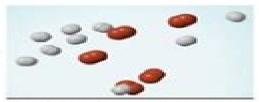
Depends on which poles of the magnets were facing each other. If one was positive and one was negative, they would be attracted to each other. If they were either both positive or negative, they would repel.

3. Think about the magnets in terms of energy. In which case do you increase the potential energy of the magnets? In which case do you increase the kinetic energy of the magnets?

Kinetic energy increases if you were to move magnets with the same charge towards each other because they are going to repel. Potential energy would increase if they were different poles because they are attracted to each other.

Gizmo Warm-up

Just like magnets, atoms of different elements are attracted together to form **chemical bonds**. Breaking these bonds requires energy. When a new bond forms, energy is released and temperatures rise. In the Reaction Energy Gizmo, you will explore how the energy of chemical bonding relates to temperature changes that occur during chemical reactions.



To begin, check that **Reaction 1** and **Forward** are selected. In this reaction, hydrogen (H) and oxygen (O<sub>2</sub>) react to form water (H<sub>2</sub>O). The reaction takes place inside a device called a **calorimeter**. Inside the calorimeter, a small chamber holds the reactants. The rest of the calorimeter is filled with water.

Click Play ( ). What happens? \_\_\_\_

Temperature increases, moving molecules.

How does the temperature change?

Increases

# **Reaction Energy Modern Chemistry Answers**

Dr. G. Mallikarjun

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