When the rates of the forward and reverse reactions are equal, the reaction has reached a state of balance called **chemical equilibrium**.

The relative concentrations of the reactants and products at equilibrium constitute the **equilibrium position** of a reaction.



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The book delves into Reaction Rates And Equilibrium Answers Prentice Hall. Reaction Rates And Equilibrium Answers Prentice Hall is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Reaction Rates And Equilibrium Answers Prentice Hall, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 4: Reaction Rates And Equilibrium Answers Prentice Hall in Specific Contexts
 - ∘ Chapter 5: Conclusion
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