



Reaction Rates and Equilibrium Study Guide

Topics Covered:

Factors affecting reaction rates
Collision theory of reaction rates
Reaction progress diagrams (activation energy and transition states)
Rate laws and the method of initial rates
Chemical equilibrium
Equilibrium expressions and constants and their meaning
Product-favored vs. Reactant-favored
Shifting equilibrium and Le Chatelier's Principle
Quantitative calculations of equilibrium systems using ICE tables

Practice Questions: [Click here for an answer key](#)

- Describe how each of the following factors would affect the rate of "particle collisions" in your response.
 - Temperature decrease
 - Addition of iodide ions, I^- , as a catalyst
 - Increased reactant concentration
 - Crushing a solid into a powder (with larger surface area)
- The reaction progress diagram shows the energy changes that take place in the reaction of carbon monoxide gas with nitrogen dioxide gas.



Reaction Rates Chemical Study Guide

Dr. Abhishek M



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Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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