

# Equilibrium and Reaction Rates

## (answers at the end)

Q: What is Le Chatelier's law (Le Chatelier's principle)?

A: Study Le Chatelier's principle's meaning.

A: Understanding that when the system is subjected to a perturbation, it reacts.

A: Study Le Chatelier's law, which predicts the direction of the reaction when the system is subjected to a perturbation (change in concentration, pressure, temperature).

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Q: What is Le Chatelier's law (Le Chatelier's principle)?

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A: Study Le Chatelier's law.

Le Chatelier's law predicts the direction of the reaction when the system is subjected to a perturbation.

- (A) The reaction is exothermic.
- (B) The reaction is endothermic.
- (C) The reaction is reversible.
- (D) It is affected by the change in concentration.
- (E) The reaction is irreversible.

A: Study Le Chatelier's law, which predicts the direction of the reaction when the system is subjected to a perturbation.

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# Reaction Rates And Equilibrium Section Review Answers

**Gordon C. Oates**



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