ASSIFICATION OF PART I **IEMICAL REACTIONS** sify the reactions below as synthesis, decomposition, single replacement or double replacement. ANSWERS Synthesis $2H_2 + O_2 \rightarrow 2H_2O$ Decomposition 2H₂O → 2H₂ + O₂ Single Replacemen $Zn + H_2SO_4 \rightarrow ZnSO_4 + H_2$ Synthesis 2CO + O₂ → 2CO₂ Decomposition 2HgO → 2Hg + O₂ Single Replacem $2KBr + Cl_2 \rightarrow 2KCl + Br_2$ Synthesis $CaO + H_2O \rightarrow Ca(OH)_2$ Double Replacemen AgNO₃ + NaCl → AgCl + NaNO₃ Decomposition 2H₂O₂ → 2H₂O + O₂

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