

DESCRIBING CHEMICAL REACTIONS

Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation.

Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical ____ 1 _____. The substances that undergo a chemical change are the ____ 2 _____. The new substances formed in a chemical reaction are the ____ 3 _____. In accordance with the law of conservation of ____ 4 _____, a chemical equation must be balanced. When balancing an equation, you place ____ 5 _____ in front of reactants and products so that the same number of atoms of each ____ 6 _____ are on each side of the equation. An equation must never be balanced by changing the ____ 7 _____ in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is ____ 8 ____; for a solid, ____ 9 ____; and for a gas, ____ 10 _____. A substance dissolved in water is designated ____ 11 _____. If a ____ 12 _____ is used to increase the rate of a chemical reaction, its formula is written above the arrow.

- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- _____ 14. The symbol Δ placed over the arrow in an equation means that heat is supplied to the reaction.
- _____ 15. Atoms are destroyed in a chemical reaction.
- _____ 16. A skeleton equation is not a balanced equation.

Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I

a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.

b. Acetylene gas (C_2H_2) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

Section 1describing Chemical Reaction Answer Key

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**



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Holt Chemistry Ralph Thomas Myers,2004 *Prentice Hall Science Explorer: Chemical Interactions* Pearson/Prentice Hall,Michael J. Padilla,2004-10 *Elements of Chemical Reaction Engineering* EduGorilla Prep Experts,2024-06-09

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Chemistry Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

The Go-To Guide for Engineering Curricula, Grades 6-8 Cary I. Sneider, 2014-11-25 How to engineer change in your middle school science classroom With the implementation of the Next Generation Science Standards your students won't just be scientists they'll be engineers But that doesn't mean you need to reinvent the wheel Respected science educator Cary Sneider has done the groundwork for you collecting a full range of time tested curriculum materials to seamlessly weave engineering and technology concepts into your math and science lessons In this volume you'll find descriptions of instructional materials specifically created for and tested in middle school science classrooms Features include A handy table that takes you straight to the chapters most relevant to your needs In depth commentaries and illustrative examples that demystify engineering curricula at the middle school level A vivid picture of what each curriculum looks like in the classroom the learning goals it accomplishes and how it helps address the NGSS More information on the integration of engineering and technology into 21st century science classrooms and why it will make a difference One of the most well respected science educators in the country Cary Sneider was an NGSS Writing Team Leader and is an associate research professor at Portland State University This publication uses hands on explorations that impact students by getting them to think like an engineer It's also great for exploring the engineering world through experiences using science and engineering and for the actual doing of science and engineering using the design process Kendall Starkweather Executive Director International Technology Education Association This book will help you engage your students in grade level engineering activities All you need to do is pick it up and get ready to implement it in your classroom

Jo Ann Vasquez Vice President Educational Practice for Helios Education Foundation CBSE Class 12 - Food Production
Question Bank Mocktime Publication,101-01-01 CBSE 12 Question Bank Food Production *Environmental Science*

Bernard J. Nebel,Richard T. Wright,1993 Revolving around the principles of sustainability this new edition sets out to provide students with a balanced complete treatment of environmental issues their scientific basis history and future Material is revised to reflect changing environmental understanding and issues **Essentials of Chemical Reaction Engineering** H. Scott Fogler,2011 Accompanying DVD ROM contains many realistic interactive simulations *Ions and Ion Pairs and Their Role in Chemical Reactions* J. Smid,2013-09-03 Ions and Ion Pairs and their Role in Chemical Reactions covers a wide spectrum of ion pairing phenomena This book focuses on the structure dynamics and energetics of ions and ion pairs and their complexes with solvent molecules and ion coordinating ligands in the solid liquid and gaseous state Organized into 12 chapters this book begins with an overview of the theory and complications involved in dipole moment measurements in slightly conductive polar solvents This text then explores the measurement of gas phase ion molecule equilibria by special spectrometric methods Other chapters consider the NMR experiments on potassium biphenyl sodium biphenyl and rubidium biphenyl dissolved in different solvents This book discusses as well the ion pair formation of some cyclic conjugated carbanions and nitranions The final chapter deals with the information obtained from relaxation measurements on processes such as ionic dissociation aggregation and solvent separation This book is a valuable resource for chemists organic chemists and quantum chemists Organic Chemistry Michael B. Smith,2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules **Chemistry** John A. Olmsted,Gregory M. Williams,Robert

Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant Microbiology Dave Wessner, Christine Dupont, Trevor Charles, Josh Neufeld, 2020-12-03 Ideal for microbiology science majors The third edition of Microbiology provides in depth coverage of the science of microscopic organisms Providing a balanced presentation of foundational concepts real world applications and current research and experimentation this comprehensive textbook facilitates a thorough understanding of the scope nature and complexity of microbiology The text approaches the subject within the context of exploration and experimentation integrating a wealth of classroom tested pedagogical features The material is organized around the three pillars of physiology ecology and genetics helping students appreciate the interconnected and dynamic nature of microbiology as they explore individual microbes and the relation between different types of microbes other organisms and the environment Detailed yet accessible chapters illustrate how an experiment proceeds explain how microbes replicate clarify the flow of concept processes and summarize key points Challenging end of chapter questions both test students understanding of the material and strengthen critical thinking skills This new edition contains up to date coverage of topics including DNA replication and gene expression viral pathogenesis microbial biotechnology adaptive immunity the control of infectious diseases the microbiology of food and water and integrated coverage of COVID 19 **Physical and Chemical Changes** Edward P. Ortleb, Richard Cadice, 1993-09-01 General chemistry information including everything from matter to radioactivity For grades 5 to 9

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