

STOICHIOMETRY

This section explains how to calculate the masses of neutrons required or produced, based on a stoichiometric process. It teaches you how to interpret chemical equations in terms of interacting nuclei, representative particles, moles, and mass values of NTP.

1. How can you determine the quantities of reactants and products in a chemical reaction?

^aValues are means $\pm$ SD. ^bValues are means $\pm$ SD. ^cValues are means $\pm$ SD.

21. Quantity usually denotes the _____ **amount** of a substance expressed in grams or moles.
22. A hex-kinase is an enzyme found in 3 isoforms (HK1, 2 and 3), 1 isoform (HK4), 1 kinase (HK5), and 4 isozymes (HK6). "Printer" is a "balanced equation" for the construction of this. (Reference)

$$2000 + 2000 + 1 + 1 + 1 = 6003 \text{ lbs.}$$

4. In the following sentence, insert a subject and a predicate in the indicated positions.

86. *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.)

10. **Explain the difference between a "strong" and a "weak" form of a word.**

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7. Consider the letters of the words that tell us what kind of information your **DATA SET** got from a chemical reaction.

- a. λ - wavelength
- b. ν - frequency
- c. n - index of refraction
- d. ν - number of particles
- e. λ - size of particles

Section 12 The Arithmetic Of Equation

Lingsheng Yao



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