



## Unit 1: Chemical Changes

# Chemical Changes

Read the passage and answer the following questions.



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) and the products are water ( $\text{H}_2\text{O}$ ) and oxygen ( $\text{O}_2$ ). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in dentistry as a bleaching agent. The reaction of hydrogen peroxide to form water and oxygen is a common example of a chemical change.

4. According to the law of conservation of mass, if the mass of the reactants is 100 grams, what is the mass of the products? Is it 100 grams, 200 grams, or 50 grams?

100 grams



5. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product?

6. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

7. What is the chemical formula for hydrogen peroxide? Is it a liquid, solid, or gas? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in dentistry as a bleaching agent.

8. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product?

9. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

In nature, the reaction of  $\text{H}_2\text{O}_2$  to form  $\text{H}_2\text{O}$  and  $\text{O}_2$  is a common example of a chemical change.

10. According to the law of conservation of mass, if the mass of the reactants is 100 grams, what is the mass of the products? Is it 100 grams, 200 grams, or 50 grams?

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**M Lipman**



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