

Chapter 2, The Chemistry of Life (continued)**Section 2–3 Carbon Compounds** (pages 44–48)

This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things.

The Chemistry of Carbon (page 44)

1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons.
2. What gives carbon the ability to form chains that are almost unlimited in length? A carbon atom can bond to other carbon atoms.

Macromolecules (page 45)

3. Many of the molecules in living cells are so large that they are known as macromolecules.
4. What is the process called by which macromolecules are formed? Polymerization
5. When monomers join together, what do they form? Polymers
6. What are four groups of organic compounds found in living things?
 - a. Carbohydrates
 - b. Lipids
 - c. Nucleic acids
 - d. Proteins

Carbohydrates (pages 45–46)

7. What atoms make up carbohydrates? Carbon, hydrogen, and oxygen atoms make up carbohydrates.
8. Circle the letter of each sentence that is true about carbohydrates.
 - ☒ a. Starches and sugars are examples of carbohydrates.
 - ☒ b. Living things use them as their main source of energy.
 - ☐ c. The monomers in sugar polymers are starch molecules.
 - ☒ d. Plants and some animals use them for strength and rigidity.
9. Single sugar molecules are also called monosaccharides.
10. Circle the letter of each monosaccharide.
☒ a. galactose b. glycogen ☒ c. glucose ☒ d. fructose
11. What are polysaccharides? They are large macromolecules formed from monosaccharides.
12. How do plants and animals store excess sugar? Plants use a polysaccharide called plant starch, whereas animals use a polysaccharide called glycogen.

Section 2 3 Carbon Compounds Pages 44 48 Answers

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