

Sample Question Paper

4

Chemistry

A sample question paper for **CBSE** Class XI

Time : 3 hrs

Max. Marks : 70

General *Instructions*

1. All questions are compulsory.
2. The question paper consists of 26 questions divided into five sections A, B, C, D and E. Section A comprises of 5 questions of 1 mark each, Section B comprises of 5 questions of 2 marks each, Section C comprises of 12 questions of 2 marks each, Section D comprises of 1 questions of 4 marks and Section E comprises of 3 questions of 5 marks each.
3. All questions in Section A are to be answered in one word, one sentence or as per the exact requirement of the question.
4. There is no overall choice. However, internal choice has been provided in some questions. You have to attempt only one of the alternative in all such questions.
5. Use of calculator is not permitted. You may ask for logarithmic table, if required.

Section A

1. Is the law of constant composition true for all types of compounds? Mention why or why not?
2. What was the necessity for the classification of the elements?
3. Water decomposes by absorbing 286.2 kJ of electrical energy per mole. When H_2 and O_2 combine to form one mole of H_2O , 286.2 kJ of heat is produced. Which law is proved? What statement of the law follows from it?
4. Which d-orbital does not have four lobes and what is its shape called?
5. Which species is the smallest aromatic substance?

Section B

6. What is hybrid state of central atom in the following?

NO_3^- , BF_4^- , PF_5 , IF_5 and CO_2

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