

Name: NATALIA ZUNIGA

Student Exploration: Solubility and Temperature

Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- What happens when you stir a spoonful of sugar into hot water?
- When sugar or another substance is **dissolved** in water, it disappears from view and forms a **homogeneous mixture** with the water, also called a **solution**.

If you can't see the sugar, how can you tell that it is there?

- Does sugar dissolve more easily in hot water or cold water?

Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the *Solubility and Temperature* Gizmo, you will study how temperature affects how much solute will dissolve in a solution.

To begin, check that **Potassium nitrate** is selected and the **Temp.** of the water is 20 °C. Click **OK**.

- In this solution, what is the solute? What is the solvent?
- Click **Add 10 g** to mix 10 g of potassium nitrate into the water.
 - Did all of the potassium nitrate dissolve?
 - How can you tell?

Activity A:

Solubility

Get the Gizmo ready:

- Click **Reset**.

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