Atomic and Ionic Radii Mr Cole Chemistry



Trends in Atomic Radii

1	Describe and explain the trend in atomic radii as you go down group 2.
2	Describe and explain the trend in atomic radii as you go across period 3.
a_	Describe and explain the trend in atomic radii as you go across period 4 from Scandium to Zinc.
314	dii Comparisons
ń	pare the radii of the following, including a description and explanation in your answer.
	a. No and Na*
-	
	b. Li and O
-	

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National Information Center for Educational Media Teaching Transparency 2atomic And Ionic Radii:

Index to Educational Overhead Transparencies National Information Center for Educational Media, 1975

Directory W/Tchng Nts Holt Chemfile Holt Rinehart & Winston, 1998-01-25

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