

Why aren't the elements listed in alphabetical order?

As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily. But to a chemist, the periodic table is a very powerful tool. The periodic table is organized by properties, both chemical and physical. Those properties relate to the electronic structure of the atoms of each element. In today's activity, you are going to study how the ground state electron configurations and structure of atoms are related to the shape and organization of the periodic table.

A blank periodic table grid with the following structure:

- Periods (Rows):** 1 to 7.
- Groups (Columns):** 1 to 18.
- Blocks:**
  - s-block:** Groups 1 and 2.
  - p-block:** Groups 13 to 18.
  - d-block:** Groups 3 to 10.
  - f-block:** Groups 14 to 17 (located below the main grid).
- Labels:**
  - s-block:** Located above the first two columns.
  - p-block:** Located above the last six columns.
  - d-block:** Located above the middle six columns.
  - f-block:** Located below the main grid, centered under groups 14-17.
- Periodic Table Symbols:**
  - Period 1: H (Group 1), He (Group 18).
  - Period 2: Li (Group 1), Be (Group 2), B (Group 13), C (Group 14), N (Group 15), O (Group 16), F (Group 17), Ne (Group 18).
  - Period 3: Na (Group 1), Mg (Group 2), Al (Group 13), Si (Group 14), P (Group 15), S (Group 16), Cl (Group 17), Ar (Group 18).
  - Period 4: K (Group 1), Ca (Group 2), Sc (Group 3), Ti (Group 4), V (Group 5), Cr (Group 6), Mn (Group 7), Fe (Group 8), Co (Group 9), Ni (Group 10), Cu (Group 11), Zn (Group 12), Ga (Group 13), Ge (Group 14), As (Group 15), Se (Group 16), Br (Group 17), Kr (Group 18).
  - Period 5: Rb (Group 1), Sr (Group 2), Y (Group 3), Zr (Group 4), Nb (Group 5), Mo (Group 6), Tc (Group 7), Ru (Group 8), Rh (Group 9), Pd (Group 10), Ag (Group 11), Cd (Group 12), In (Group 13), Sn (Group 14), Sb (Group 15), Te (Group 16), I (Group 17), Xe (Group 18).
  - Period 6: Cs (Group 1), Ba (Group 2), La (Group 3), Ce (Group 4), Pr (Group 5), Nd (Group 6), Pm (Group 7), Sm (Group 8), Eu (Group 9), Gd (Group 10), Tb (Group 11), Dy (Group 12), Ho (Group 13), Er (Group 14), Tm (Group 15), Yb (Group 16), Lu (Group 17), Hf (Group 18), Ta (Group 19), W (Group 20), Re (Group 21), Os (Group 22), Ir (Group 23), Pt (Group 24), Au (Group 25), Hg (Group 26), Tl (Group 27), Pb (Group 28), Bi (Group 29), Po (Group 30), At (Group 31), Rn (Group 32).
  - Period 7: Fr (Group 1), Ra (Group 2), Ac (Group 3), Th (Group 4), Pa (Group 5), U (Group 6), Np (Group 7), Pu (Group 8), Am (Group 9), Cm (Group 10), Bk (Group 11), Cf (Group 12), Es (Group 13), Fm (Group 14), Md (Group 15), No (Group 16), Lr (Group 17), Rf (Group 18), Db (Group 19), Sg (Group 20), Bh (Group 21), Hs (Group 22), Mt (Group 23), Ds (Group 24), Rg (Group 25), Co (Group 26), Bg (Group 27), Hg (Group 28), Ts (Group 29), Og (Group 30).

- ## Cracking the Periodic Table Code

# Teacher Guide Cracking The Periodic Table Code

**TD Snyder**



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