



Trends in Atomic Radii

1. Describe and explain the trend in atomic radii as you go down group 2.

2. Describe and explain the trend in atomic radii as you go across period 3.

3. Describe and explain the trend in atomic radii as you go across period 4 from Scandium to Zinc.

Radii Comparisons

Compare the radii of the following, including a description and explanation in your answer.

- a. Na and Na⁺ _____

- b. Li and O _____

Teaching Transparency Atomic And Ionic Radii

**Society of glass technology, Sheffield,
Eng**



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