

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- A**: Top left corner.
- B**: Top right corner.
- C**: Second row from top, second column from left.
- D**: Second row from top, eighth column from left.
- E**: Second row from top, tenth column from left.
- F**: Third row from top, second column from left.
- G**: Third row from top, sixth column from left.
- H**: Third row from top, eighth column from left.
- I**: Fourth row from top, tenth column from left.
- J**: Fifth row from top, second column from left.
- K**: Fifth row from top, twelfth column from left.
- L**: Fifth row from top, third column from left.
- M**: Fifth row from top, sixth column from left.
- N**: Fifth row from top, eighth column from left.
- O**: Fifth row from top, tenth column from left.
- P**: Fifth row from top, twelfth column from left.
- Q**: Sixth row from top, second column from left.
- R**: Sixth row from top, third column from left.
- S**: Sixth row from top, sixth column from left.
- T**: Sixth row from top, seventh column from left.
- U**: Sixth row from top, eighth column from left.
- V**: Seventh row from top, second column from left.
- W**: Seventh row from top, eighth column from left.
- X**: Eighth row from top, second column from left.
- Y**: Eighth row from top, eighth column from left.
- Z**: Eighth row from top, sixth column from left.

Arrows point from the letters **R** and **Z** to a separate section below the grid, which contains the letters **V**, **W**, **X**, and **Y** in a grid structure.

List elements in *alphabetical order* when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Teaching Transparency Worksheet 18 The Periodic Table Answer Key

SJ Ball



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Glencoe Science McGraw-Hill Staff, 2001-06

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