

# The Periodic Table and Periodic Law

## Section 6.1 Development of the Modern Periodic Table

In your textbook, read about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

|          |              |                  |          |
|----------|--------------|------------------|----------|
| octaves  | atomic mass  | atomic number    | nine     |
| elements | properties   | Henry Moseley    | eight    |
| protons  | periodic law | Dmitri Mendeleev | accepted |

The table below was developed by John Newlands and is based on a relationship called the law of (1) \_\_\_\_\_. According to this law, the properties of the elements repeated every (2) \_\_\_\_\_ elements. Thus, for example, element two and element (3) \_\_\_\_\_ have similar properties. The law of octaves did not work for all the known elements and was not generally (4) \_\_\_\_\_.

|   |    |    |    |    |    |    |
|---|----|----|----|----|----|----|
| 1 | 2  | 3  | 4  | 5  | 6  | 7  |
| H | Li | G  | Bo | C  | N  | O  |
| 8 | 9  | 10 | 11 | 12 | 13 | 14 |
| F | Na | Mg | Al | Si | P  | S  |

The first periodic table is mostly credited to (5) \_\_\_\_\_. In his table, the elements were arranged according to increasing (6) \_\_\_\_\_. One important result of this table was that the existence and properties of undiscovered (7) \_\_\_\_\_ could be predicted.

The element in the modern periodic table are arranged according to increasing (8) \_\_\_\_\_, as a result of the work of (9) \_\_\_\_\_. This arrangement is based on number of (10) \_\_\_\_\_ in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) \_\_\_\_\_, which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) \_\_\_\_\_.

# Teaching Transparency Worksheet 18 The Periodic Table Answers

**George Graybill**



## **Teaching Transparency Worksheet 18 The Periodic Table Answers:**

**Glencoe Science** McGraw-Hill Staff,2001-06      **Atoms, Molecules & Elements: What Are Compounds? Gr. 5-8**

George Graybill,2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

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