

Name: _____

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States of Matter Test

1. Which of the following changes made to a sample of liquid would result in an increase in the distance between the individual molecules of the substance?
- Transferring the substance to a smaller container
 - Adding heat energy to the substance
 - Changing the phase of matter from a liquid to a solid
 - Decreasing the overall kinetic energy (motion) of the liquid
2. A student drew a representation of the atoms of a solid. The illustration shows closely packed atoms vibrating, but remaining in the same relative position.

Which of the following changes must the student make to the illustration to accurately

represent the same substance in the liquid phase?

- The illustration should show increased movement of the atoms
 - Each individual atom should be much larger
 - There should be very wide spaces between the atoms
 - The shape of each atom must reflect the shape of its container?
3. As thermal energy is added to a substance, which of the following changes can be predicted?
- Spacing between particles will decrease

Test Unit 13 Matter

Rachel Sandford



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