

## Practice Final Exam

Pick the best answer for each of the following:

\_\_\_\_\_ 1. Which of the following is NOT a true statement concerning what happens in a chemical reaction?

- The ways in which atoms are joined together are changed
- New atoms are formed as products
- The starting materials are called reactants
- The bonds of the reactants are broken and new bonds of the products are formed
- In a word equation representing a chemical reaction, the reactants are written on the left and the products on the right.

\_\_\_\_\_ 2. In the balanced chemical equation for the following reaction, what is the coefficient in front of fluorine?



a) 1    b) 2    c) 3    d) 5    e) 6

\_\_\_\_\_ 3. In order to predict whether or not a redox (oxidation-reduction) reaction takes place, we

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a) measure electron configuration  
b) orbital energy level diagram

\_\_\_\_\_ 4. Predict the products for the following reaction:



a) potassium zinc + nitrogen  
b) potassium nitride + zinc  
c) potassium nitrate + zinc  
d) potassium nitrate + zinc  
e) No Reaction

Lithium	<div style="display: inline-block; width: 10px; height: 40px; background-color: black; position: relative; top: -5px;"> <span style="position: absolute; top: -10px; right: -5px; transform: rotate(90deg); font-size: 8px;">Decreases</span> </div>
Potassium	
Calcium	
Magnesium	
Zinc	

\_\_\_\_\_ 5. Which of the following accurately represents the correct balanced equation for the following reaction:



a)  $5 \text{ S} + \text{N} \rightarrow \text{N}_5$   
b)  $5 \text{ S} + \text{N} \rightarrow \text{N}_5\text{S}$   
c)  $3 \text{ S}_8 + \text{N}_2 \rightarrow \text{N}_2\text{S}_6$   
d)  $6 \text{ N}_2 + \text{S}_8 \rightarrow 2 \text{ N}_2\text{S}_6$   
e)  $4 \text{ S}_8 + \text{N}_2 \rightarrow 2 \text{ N}_2\text{S}_8$

# Texas Grade 1chemistry Final Practice Exam

**BM King**



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