

Organic Chemistry I Assessment Exam Answer Key  
Created by Melissa Maribel

1. How many core electrons and valence electrons does each have?

a. O and S

Both have  
6 valence electrons

b. C and Si

Both have  
4 valence electrons

c. N and P

Both have  
5 valence electrons

2. Write the full electron configuration for each atom:

a. O

$1s^2 2s^2 2p^4$

b.  $O^-$

$1s^2 2s^2 2p^5$

c. N

$1s^2 2s^2 2p^3$

d.  $N^+$

$1s^2 2s^2 2p^2$

3. For the compound methanol,  $CH_3OH$ , which of the following statements is true for the oxygen atom? (Circle one)

- ☒ a. It has no formal charge, but has a slight negative charge.
- ☐ b. It has no formal charge, but has a slight positive charge.
- ☐ c. It has a formal +1 charge.
- ☐ d. It has a formal -1 charge.



# Test 1 The Arrangement Of Electrons Answer Key

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