



$$\text{Change in } E = q + w$$

$$w = -p \Delta V$$

- q = heat transfer
- W = work

$$w = -(5.0 \text{ atm} (3.0 \text{ L} - 2.0 \text{ L}))$$

$$101.3 \frac{\text{J}}{\text{atm} \cdot \text{L}}$$

$$-5.0 \text{ atm} \times 1.0 \text{ L} = -5.0 \text{ atm} \cdot \text{L}$$

- Example

Calculate the change in E . 300.0 J was absorbed by a system and the gas expanded from 2.0 L to 3.0 L at a pressure of 5.0 atm

$$\Delta E = (300.0 \text{ J}) +$$

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Mahesh Dattani



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