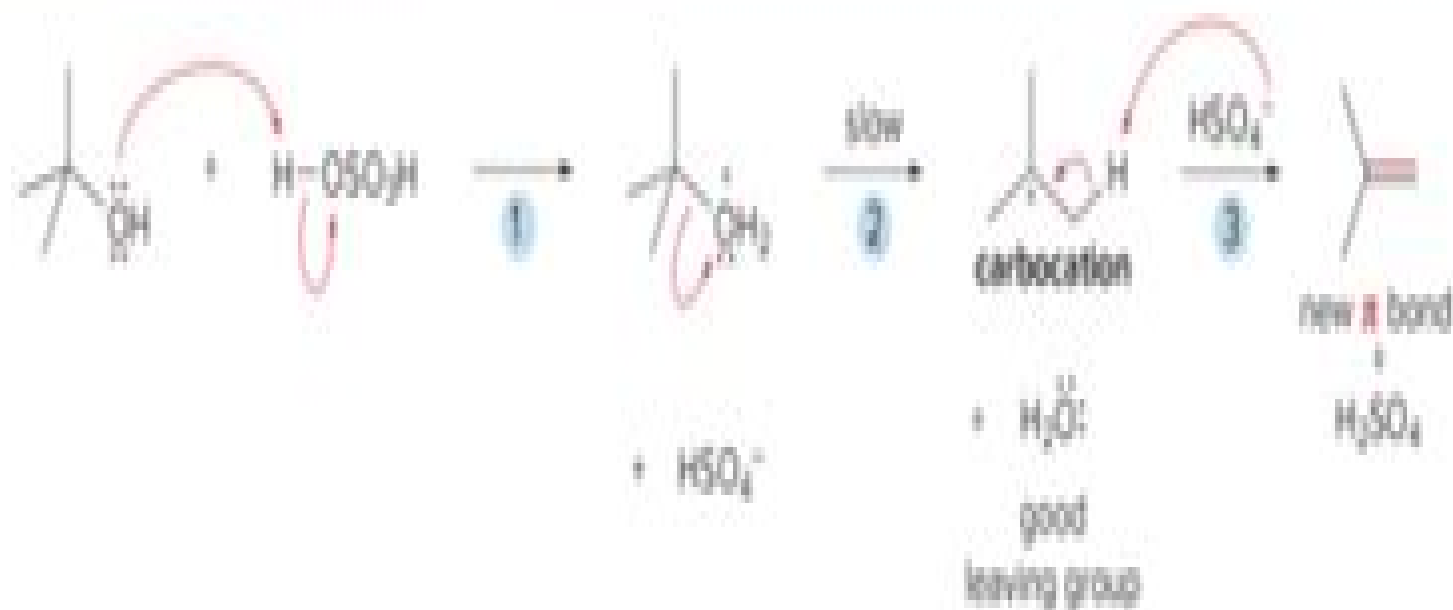




## Mechanism 9.1 Dehydration of 2° and 3° ROH—An E1 Mechanism



- 1 Protonation of the oxygen atom converts the poor leaving group ( $^- \text{OH}$ ) into a good leaving group ( $\text{H}_2\text{O}$ ).
- 2 Heterolysis of the  $\text{C}-\text{O}$  bond forms a **carbocation** in the rate-determining step.
- 3 A base (such as  $\text{HSO}_4^-$  or  $\text{H}_2\text{O}$ ) removes a proton from a carbon adjacent to the carbocation to form the new  $\pi$  bond.

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