

Test 7: Periodic Table Review Questions

Name: \_\_\_\_\_

Wednesday, January 16, 2008

1.

Which halogen is a solid at STP?

- 1. fluorine      3. bromine
- 2. chlorine     4. iodine

2.

Element *M* is a metal and its chloride has the formula  $MCl_2$ . To which group of the Periodic Table does element *M* most likely belong?

- 1. 1 (IA)      3. 15 (VA)
- 2. 2 (IIA)     4. 17 (VIIA)

3.

As the atomic number of elements within Group 2 increases, the metallic character of each successive element

- 1. decreases
- 2. increases
- 3. remains the same

4.

Which characteristics describe most nonmetals in the solid phase?

- 1. They are malleable and have metallic luster.
- 2. They are malleable and lack metallic luster.
- 3. They are brittle and have metallic luster.
- 4. They are brittle and lack metallic luster.

5.

Which group in the Periodic Table contains elements that are all gases at STP?

- 1. 11 (IB)      3. 12 (IIB)
- 2. 17 (VIIA)   4. 18 (0)

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**The Princeton Review**



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