

## Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
  - a) **White**
  - b) Cream
  - c) Yellow
  - d) Green
2. What colour do sodium ions go in a flame test?
  - a) Red
  - b) **Yellow**
  - c) Brick-red
  - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
  - a) Silver chloride and nitric acid
  - b) Barium chloride and hydrochloric acid
  - c) **Silver nitrate and nitric acid**
  - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
  - a) Lit splint goes out with a squeaky pop
  - b) Damp blue litmus paper goes red
  - c) Damp red litmus paper goes blue
  - d) **Glowing splint relights**
5. What colour precipitate do iodide ions form?
  - a) White
  - b) Cream
  - c) **Yellow**
  - d) Green
6. Which ion produces a brick-red colour in a flame test?
  - a) Sodium
  - b) **Calcium**
  - c) Copper
  - d) Lithium
7. What is the test for carbon dioxide gas?
  - a) **Limewater turns cloudy**
  - b) Damp blue litmus turns red
  - c) A glowing splint relights
  - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
  - a) White
  - b) Blue
  - c) Green
  - d) **Red/brown**
9. What is the test for chlorine gas?
  - a) **Damp blue litmus paper turns red and is then bleached white**
  - b) Red litmus ☐ turns blue
  - c) A glowing splint relights
  - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
  - a) **Blue**
  - b) Green
  - c) Red/brown
  - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
  - a) Silver chloride and nitric acid
  - b) Barium chloride and sulphuric acid
  - c) Silver nitrate and nitric acid
  - d) **Barium chloride and hydrochloric acid**
12. What colour do potassium ions go in a flame test?
  - a) **Lilac**
  - b) Red
  - c) Yellow
  - d) Brick-red
13. What is the test for hydrogen gas?
  - a) **Lit splint goes out with a squeaky pop**
  - b) Glowing splint relights
  - c) Damp blue litmus paper turns red and then is bleached white
  - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
  - a) Sodium
  - b) Calcium
  - c) Copper
  - d) **Lithium**
15. What colour precipitate do bromide ions form?
  - a) White
  - b) **Cream**
  - c) Yellow
  - d) Green
16. What needs to be added to test for the presence of carbonate ions?
  - a) Silver nitrate
  - b) Universal indicator
  - c) Litmus paper
  - d) **Dilute acid**
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
  - a) Oxygen
  - b) **Ammonia**
  - c) Carbon dioxide
  - d) Hydrogen
18. What colour precipitate does barium sulphate form?
  - a) **White**
  - b) Green
  - c) Yellow
  - d) Cream
19. What is needed to carry out a flame test?
  - a) Splint
  - b) Glass rod
  - c) Spatula
  - d) **Nichrome/platinum wire**

Your score out of 19=  
Percentage =

# Test Solutions Answer Key Chemistry

**Lauren Gardner**



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