1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Conducts electricity
Contains hydrogen ion
Proton donor
Turns litmus paper red
React with most metals
Releases hydronium
Ex: Jemon
Witherstar

Battery acid (sulfuric acid)

Action

Distribute:

conducts electricity contains hydroxide ion proton acceptor turns litmus paper blue releases hydroxide ion. Ex: Windex blench SCHOOL ST

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, acids or bases:

20.3	A.	produce hydrogen ions (H') in solution
b)	88	have a bitter taste
e) _	ER.	produce hydroxide ions (OH') ions
A876	100	property of the State of the Contract of the C

e) ____ A ____ have a sour taste

f) B have a slippery, soapy feel

g) ____ B ____ react with acids to form salts

h) A react with many metals

3. Write an equation for the dissociation of the following acids in water: follow example

d)
$$H_3S \rightarrow 2H^2 + S^2$$

4. Write an equation for the dissociation (ionization) of the following bases in water: (follow example)

5) A. What type of reaction is a neutralization reaction? Double replacement

B. Write the net equation for neutralization:

$$H^+ + OH^- \rightarrow H_2O$$

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S Ashworth

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