National Co.	Theter:	
	L. Carlotte	

9-3 Review and Reinforcement

Classifying chemical Reactions

On the line as the light wides the letter of the type of chamical reaction represented by each numerion before.

1. $A + B \rightarrow AB$

a, decomposition reaction.

 $_{---}$ 2. A + BX \rightarrow AX + B

4. All + A + B

by direct combination reaction

- $X = X + BY \Rightarrow AY + BX$
- double-replacement reaction
 d. single-replacement reaction

Describe in words each of the following types of reactions.

- 5. dineat combination reaction
- 6. decomposition reaction
- T. stagle-replacement reaction.
- N. double-replacement sourcion

Decide whether each of the following equations represents a direct combination (dc), decomposition (d), single-replacement (xr), or double-replacement (dr) reaction. Write year univers on the line.

- $\underline{}$ 9. $CO_2 \rightarrow C + O_2$
- _____ ID. NaCl + AgNO₂ → NaNO₂ + AgCl
- $11.5_0 + 401 \rightarrow 8800_0$
- 12. BaCl₂ + 2NaOH → 2NaCl + BaHOH)₂
- _____ 13. Zn + CuSO₁ → ZnSO₁ + Cu
- _____ 14, CH₀ \Rightarrow C + 2H₁
- ______15. PbrNO₁u + Mg → Pb + MgtNO₁b

Chairpisters.

Review And Reinforcement Chemical Equilibrium Answer Key

United States. National Bureau of Standards

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Nuclear Science Abstracts ,1973

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